

Tài liệu này được dịch sang tiếng việt bởi:



Từ bản gốc:

https://drive.google.com/folderview?id=0B4rAPqlxIMRDflBVQnk2SHNlbkR6NHJi $\underline{N1Z3N2VBaFJpbnlmbjhqQ3RSc011bnRwbUxsczA\&usp=sharing}$

Liên hệ dịch tài liệu:

thanhlam1910_2006@yahoo.com hoặc frbwrthes@gmail.com hoặc số 0168 8557 403 (gặp Lâm)

Tìm hiểu về dịch vụ: http://www.mientayvn.com/dich tieng anh chuyen nghanh.html

Chiral	phosphinothiourea			Chất	χί	íc	tá	
organocata	alyst	in		the	phos	phino	thio	ure
enantiosel	ective	\mathbf{N}	Ior	ita-	các	phản	ứng	N
Baylis-Hil	llman	reaction	ns	of	Hillr	nan (chọn	1
aromatic	aldel	nydes	W	ith	của	các	and	leh

hữu íс ea chiral trong Morita- Baylisloc đối quang nit thơm với methyl vinyl ketone

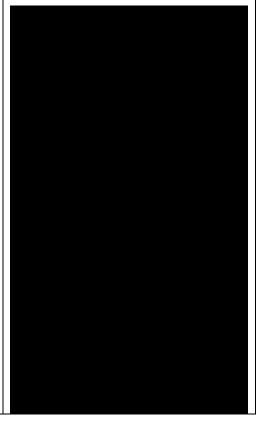
The Morita-Baylis-Hillman (MBH) reaction is a powerful tool for the atom-economic construction of densely functionlized a- methylene-phydroxycarbonyl derivatives which serve as valuable building blocks in organic synthesis.1 The development of a suitable asymmetric version of this reaction has attracted considerable interest in recent years.2 Various chiral organocatalysts including quinidine-derived pisocupreidine,3 BINOLderived Brensted acid,4 and bifunctional aminothiourea5 have been notably developed for the asymmetric MBH to reaction achieve high enantiomeric excesses.

Enantiomer còn có nghĩa là isome quang

Bis(thio)ureas derived from chiral trans-1,2-diaminocyclohexane, isophorone-diamine (IPDA), and H8-BINAM were also proven to be efficient organocatalysts for the asymmetric MBH reactions of various aldehydes with cyclohexen-1-one as described by Nagasawa, Berkessel and Shi, respectively.6 However, Morita-Baylis-Hillman reactions involving methyl vinyl ketone (MVK) reaction partner usually afforded the corresponding in moderate products enantioselectivities.2,7 To the best of our knowledge, the

methyl vinyl ketone

Phản ứng Morita-Baylis-Hillman (MBH) là một công cụ mạnh để xây dựng hiệu quả nguyên tử của các dẫn xuất amethylene-p-hydroxycarbonyl chức hóa đậm đặc, những chất này đóng vai trò là các thành phần cơ bản có giá trị trong quá trình tổng hợp hữu cơ. Sư phát triển của phiên bản bất đối xứng phù hợp của phản ứng này đã thu hút rất nhiều sự quan tâm của các nhà nghiên cứu trong những năm gần đây. Các chất hữu cơ chiral khác bao gồm p-isocupreidine có nguồn gốc từ quinidine,3 axit Brensted có nguồn gốc từ BINOL.4 và aminothiourea nhi chức cũng đã được thiết kế riêng cho phản ứng MBH bất đối xứng để cho ra lương dư đối quang cao.

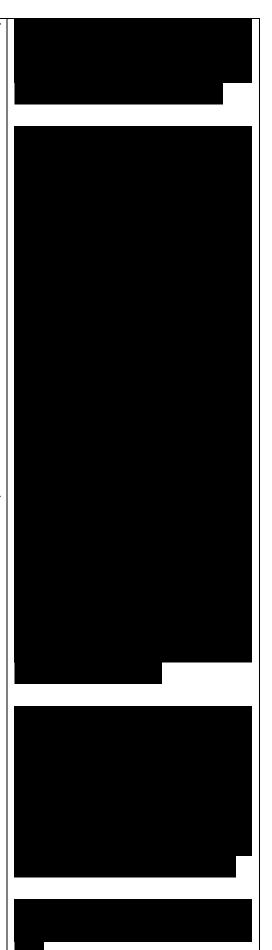


highest enantioselectivity of this kind of MBH reaction was 83% ee using chiral tertiary amine/proline cocatalysts.7d

Cyclohexane-based aminothiourea has been extensively studied in recent a bifunctional vears as organocatalyst for asymmetric catalysis.8 In 2004, Yudin and co-workers reported cyclohexanebased iminophosphines derived from the ring-opening of aziridine for transition metal catalysis.9 Due to the structural analogy between aminothiourea and iminophosphine, we envisioned that the overlapped structure. that is. phosphinothiourea (Fig. could serve as a new class of bifunctional organocatalysts for enantiose- lective catalysis. As tertiary phosphines are effective catalysts for the MBH it is reaction, conceivable that the phosphinothiourea could catalyze this reaction in an enantioselective fashion.

The bifunctional phosphinothiourea was not before documented for asymmetric catalysis until Shi and Shi10 recently reported the BINOL-derived phosphinothioureas as organocatalysts in the

Figure 1. Overlap approach to design new catalyst.



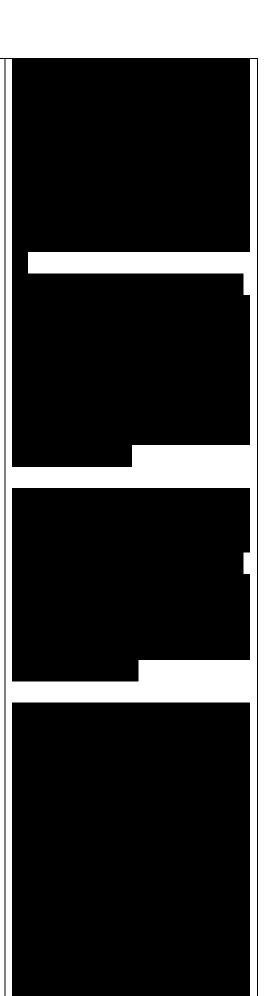
presence of carboxylic acid as co-catalyst for the asymmetric aza-Morita-Baylis-Hillman However. reaction. the reaction usually takes a long time. Herein, we report a new class of phosphinothiourea organocatalysts, derived from trans-2-amino-(diphenylphosphino)cyclohexa ne (1),11 which are fairly effective bifunctional organocatalysts for the enantioselective **MBH** reaction of arylaldehydes with **MVK** to give the corresponding adducts in up to 94% ee and good to excellent yields.

The organocatalysts 2a-f12 are easily prepared by condensation of (R,R)-2-amino-1-

(diphenylphosphino)cyclohexa ne (1) with 1.1 equiv of the corresponding

iso(thio)cyanate or isocyanate under mild conditions (Fig. 2, see Supplementary data).

We initially chose the reaction of MVK with p-nitrobenzaldehyde to screen the phosphinothioureas as bifunctional organocat- alysts for the MBH reaction. To our delight, when using CH2Cl2 solvent, the reaction as completed in 45 min with catalyst 2a to provide the desired MBH product in 71% yield and 85% ee (Table 1, entry 3). Longer reaction time afforded lower chemical yield



due

Table 1

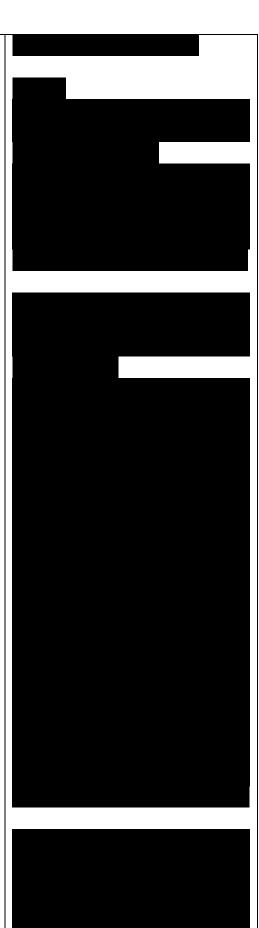
M) at 13 °C.

Screening of the catalysts for the reaction of MVK and 4nitrobenzaldehydea a Unless stated otherwise, the reactions were conducted with 10 mol % of organocatalyst, 5 equiv of MVK in CH2Cl2 (0.2

b Isolated yields. c Determined by chiral HPLC. d 5 mol % 2a was used. e Not determined.

to the over reaction with MVK (entries 4 and 5 vs 3). The thiourea moiety proved to be critical for the MBH reaction in terms of both yield and enantioselectivity. Urea derivative 2b was ineffective under identical conditions (entry 7). Surprisingly, 2c gave catalyst poor enantioselectivity, albeit with yield (entry high 8). Substitution at the 4-position of phenyl has no obvious effect on the enantioselectivity (entries 3, 9, and 10). Bearing an alkyl group, the phosphinothiourea 2f of afforded good leve1 enantioselectivity, but low yield (entry 11).

Further optimization including various solvents, temperature, and substrate concentration in the presence of catalyst 2a led to the optimal reaction conditions (see Supplementary data): using 2a as catalyst, the



reaction of 5 equiv of MVK with 0.3 M of p-nitrobenzaldehyde in CHCl3 completed in 15 min at 13 °C to provide the desired product in 75% yield and 94% ee.

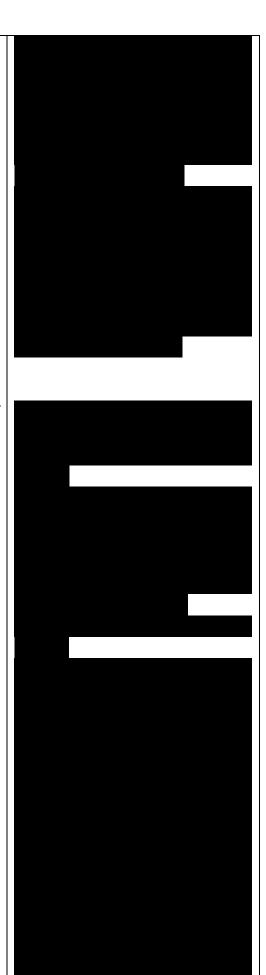
Under the optimized conditions, the substrate scope in terms of aromatic aldehydes was first investigated. As indicated in Table 2, reaction of MVK with strong electron-deficient aldehydes usually proceeded very quickly (within 1 h) to provide the desired

organocatalyst, 5 equiv of MVK in CHG3 (0.3 M) at 13 °C. b Isolated yields.

c The ee was determined by chiral HPLC, and the absolute configuration was determined by comparison of optical rotation with that of literature report.7c,13

Figure 3. Proposed transition state.

products in very good yields excellent and enantioselectivities (90-94% ee, entries 1-5). Both monodi-halogen substituted and arylaldehydes are tolerated to generate the products in high enantioselectivities (87-92% ee) albeit with moderate yields (entries 6-9). The reactions non-substituted with benzaldehyde and 2naphthylaldehyde proceeded slowly to afford the product in high enantioselecitvity (90%



ee), but low yields (entries 10 and 11). As a general trend, the reaction with electron-rich arylaldehyde proved to be difficult and complex mixture was observed.

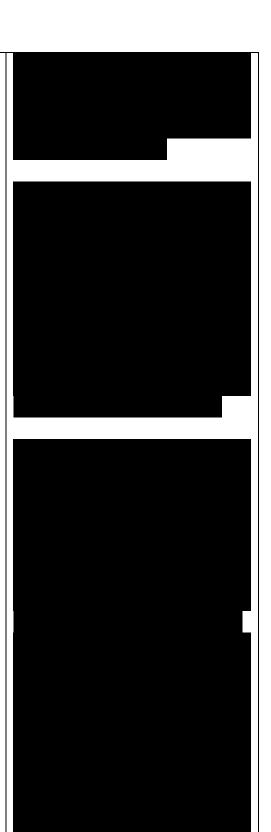
The observed absolute configuration could be explained by the plausible transition state illustrated in Figure 3. The thiourea moiety forms hydrogen-bond with the aldehyde carbonyl. The cyclohexyl scaffold forces phosphinoyl associated enolate to attack the activated carbonyl from the si-face to generate the R configuration.

In summary, have we developed a highly enantioselective Morita-Baylis-Hillman reaction addition of involving the MVK to aromatic aldehydes catalyzed by a phosphinothiourea

organocata- lyst derived from trans-2-amino-1-

(diphenylphosphino)cyclohexane. The reaction proceeds under very mild conditions to afford the desired product in a short time period in good to excellent yields with generally excellent enantiomeric excesses (87-94% ee). Further efforts are underway with a focus on improving the catalyst activity and the scope of substrate as well as the mechanism of the asymmetric MBH reaction.

Enantiomer còn có nghĩa là isome quang



•		